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**(84) ELECTRIC DOUBLE LAYER CAPACITOR.**

(87) Electric double layer capacitor having an aluminum layer (8) formed on the inner surface of at least the portion of a metal case (9) which constitutes an anode. The capacitor is formed so as to withstand voltages not less than 3V, by skillfully utilizing the characteristics of this aluminum layer (8) that it forms an electrochemically stable anodic oxide film in accordance with an applied voltage, the resistance of which film is in a low, practically negligible level.

EP 0 207 167 A1

# SPECIFICATION

## TITLE OF THE INVENTION

ELECTRIC DOUBLE LAYER CAPACITOR

## TECHNICAL FIELD

This invention relates to an electric double layer capacitor utilizing an electric double layer formed at the interface between polarizable electrodes and an electrolyte.

## BACKGROUND TECHNOLOGY

Known electric double layer capacitors of the type mentioned above are comprised of, as shown in Fig. 1, polarizable electrodes 1 which are obtained by press molding of active carbon particles, by applying a mixture of active carbon particles and an appropriate binder onto a collector metal, or by forming a spray coating layer of aluminium on active carbon fibers. The polarizable electrodes 1 are accommodated in a stainless steel case 2 and are facing each other through an electrolytic solution and a separator 3. The metallic case 2 is sealed through a gasket 4 at a peripheral opening thereof.

Another type of known capacitor now in use is shown in Fig. 2 in which a non-polarizable electrode 5 is used as one electrode.

In these known arrangements, solvents used for the electrolytic solution are propylene carbonate,  $\gamma$ -butyrolactone, N,N-dimethylformaldehyde, and acetonitrile

In the case of polarization at the anode, the stainless steel in the electrolytic solution cannot make a complete passive state but dissolves in the solution. The potential at which the current starts to run owing to the dissolution is 2.3 - 2.4 volts which is determined depending on the decomposition potential of the solvent at the cathode. This is lower than the potential of oxidation of active carbon or the potential of decomposition of the electrolyte in the electrolytic solution using the organic solvents.

Accordingly, when the stainless steel case 2 is used as a current collector, the potential at the anode is restricted by the potential of dissolution of the stainless steel, so that a potential of 3 V which is in an electrochemically stable potential region determined by the polarizable electrode 1 and the electrolytic solution could not be effectively utilized.

For instance, when an excess voltage over a voltage at which a leakage current starts to increase is applied, large amounts of iron, nickel and the like are detected in the electrode at the side of the anode, from which it has been confirmed that the stainless steel is dissolved and iron ions are moved from the anode toward the cathode.

As will be seen from the above, when stainless steel is used as the metallic case 2, it is difficult to effectively utilize a potential of 3 V which is in an electrochemically

stable region determined by the active carbon polarizable electrode 1 and the electrolytic solution. In order to obtain an electric double layer capacitor of a breakdown voltage which enables one to use 3 V, it is necessary to use a material which allows passage of a reactive current at a potential equal to or larger than active carbon of the polarizable electrode 1 and which has strength sufficient as the case if the anode is subjected to polarization in a solvent used.

One such a material may be titanium which forms a passive state in electrolytic solutions. As shown in Fig. 3, however, the breakdown voltage becomes higher than in the case using stainless steel. With electrolytic solutions using propylene carbonate and tetraethylammonium perchlorate, the region where the reactive current flows increases by about 0.8 V. However, the internal resistance increases, so that when the voltage drop becomes large in case where the electric double layer capacitor is used, with an attendant problem that such a capacitor cannot be in use. The present invention is contemplated to solve the above problem and has for its object the provision of an electric double layer capacitor which has a high breakdown voltage over 3 V.

#### DISCLOSURE OF THE INVENTION

The present invention contemplates to solve the above

problem and has for its object the provision of an electric double layer capacitor having a high breakdown voltage over 3 V.

More particularly, the present invention has such an arrangement that a metallic case piece at the side of an anode which is in contact with a conductive electrode and electrolyte is provided with an aluminium layer on the inner surface thereof. When aluminium is formed on the surface in contact with the electrolyte, an oxide film is formed on the aluminium layer according to an applied voltage. In a region to which a voltage has once applied, no reactive current passes, so that the reaction of dissolution can be inhibited. Thus, there can be obtained an electric double layer capacitor which is electrochemically stable even when a potential of 3 V is applied.

Because the aluminium oxide film thickness is small and thus the resistance is low, an increase of the internal resistance as in the case using titanium does not occur at a low voltage of about 3 volts. In this case, the collector metal serves also as a material for the case and should have high strength sufficient for the case. If the case is constituted of aluminium alone, the strength is not satisfactory. In addition, because a limitation is placed on the thickness of the case from the standpoint of a product size, too large a thickness is not practical. Under

these restricting conditions, it is necessary to use a material, such as stainless steel, which has little problem in electric connection on use as an external terminal and high strength, in combination with aluminium.

#### BRIEF DESCRIPTION OF THE DRAWINGS

Figs. 1 and 2 are, respectively, sectional views of known electric double layer capacitors; Fig. 3 is a potential-current characteristic of electric double layer capacitors using stainless steel and titanium; Fig. 4 is a sectional view of an electric double layer capacitor according to one embodiment of the invention; and Figs. 5 - 7 are, respectively, sectional views of further embodiments of the invention.

#### BEST MODE FOR CARRYING OUT THE INVENTION

Embodiments of the invention are described with reference to Figs. 4 through 7.

An embodiment shown in Fig. 4 includes polarizable electrodes 6 each made of an active carbon fiber cloth or a molding of a mixture of an active carbon powder and a binder. An aluminium conductive electrode 7 is formed on one side of the electrodes 6 by plasma spray coating. The polarizable electrodes 6 are encased in metallic case pieces 9, 10 of stainless steel which have an aluminium layer 8 on the inner surfaces thereof in such a way that the conductive electrodes 7 contact the inner surfaces of the

case pieces 9,10. The conductive electrodes 7 are connected to the metallic case pieces 9, 10 by spot welding. One of the polarizable electrodes 6 which is a counter electrode at the side of the cathode is impregnated with an electrolytic solution which has 10 wt% of tetraethylammonium tetrafluoroborate added to propylene carbonate. An ion-permeable separator 11 is provided between the polarizable electrodes 6 and a gasket 12 is arranged around the opened periphery of the metallic case pieces 9, 10 and the metallic piece 10 is subjected to curling to complete a sealed case.

A capacitor shown in Fig. 5 is an embodiment where no aluminium layer is formed on the inner surface of the metallic case piece 10 at the side of the anode.

In Fig. 6, there is shown a capacitor which is similar to Fig. 4 but a non-polarizable counter electrode 13 of, for example, lithium is used as a cathode and the metallic case pieces 9, 10 have, respectively, aluminium layers 8 on the inner surfaces thereof.

Fig. 7 shows another embodiment in which a non-polarizable electrode is used as the counter electrode 13 and the metallic case piece 10 has an inner aluminium layer at the side of the counter electrode 13.

The metallic case pieces 9, 10 may be made of, aside from stainless steel, iron, nickel, titanium and copper

alloys.

The present invention is described by way of examples.

[Example 1]

In the embodiments shown in Figs. 4 and 5, a 250 micrometer thick aluminium conductive electrode 7 is formed by plasma spray coating on one surface of a polarizable electrode 6 consisting of a phenolic active carbon fiber cloth (thickness 0.5 mm, specific surface area  $2000 \text{ m}^2/\text{g}$ ). This double-layered construction is punched in the form of a disk having a diameter of 2 cm to obtain electrodes. The electrodes are impregnated with an electrolytic solution having 10 wt% of tetraethylammonium tetrafluoroborate in propylene carbonate, after which they are superposed through separator 11. This unit is encased in a stainless steel case made of pieces 9, 10 which is covered with an aluminium layer 8 (such as a layer having a purity of 99.86% and a thickness of 70 micrometers or a layer having a purity of 99.99% and a thickness of 60 micrometers) only on the inner surface contacting the anode and on inner surfaces contacting both electrodes. A gasket 12 is provided in an opening end between the metallic case pieces 9, 10 and caulked to close the opening.

Several characteristics of the electric double layer capacitors according to the invention are indicated in Table 1 as Nos. 1 - 3. In Table 1, characteristics of an electric



double layer capacitor for comparison in which the stainless steel case is not covered with aluminium on the inner surfaces thereof are indicated as No. 6.

[Example 2]

Coconut shell active carbon particles are mixed with a polyflon binder and molded (thickness 0.5 mm, specific surface area  $800 \text{ m}^2/\text{g}$ ) to obtain a polarizable electrode 6. The electrode is formed with a 250 micrometer thick aluminium conductive electrode 7 by plasma spray coating. This double-layered construction is punched into disks having a diameter of 2 cm to obtain electrodes. The electrodes are impregnated with an electrolytic solution of 10 wt% of tetraethylammonium tetrafluoroborate in propylene carbonate and superposed through a separator 11 and sandwiched between metallic case pieces 9, 10 in which the stainless steel case piece 9 alone or pieces 9, 10 are covered with an aluminium layer 8 (purity 99.86%, thickness 70 micrometers) on the inner surface or surfaces thereof. The metallic case pieces 9, 10 are closed with a gasket 12 at the opening therebetween.

In Table 1, the characteristics of the electric double layer capacitors of the invention are shown as Nos. 4 and 5.

Table 1

No.	Case Construction	Active Carbon	Capacitance	Internal Resistance	Break-down Voltage	Life under high temperature & loading conditions
						3.0 V, 70°C 1000 hours
Capacitors of Invention:						(Variation in capacitance)
1	anode: Al-covered stainless steel (Al purity: 99.86%)	phenolic active carbon fibers	3 F	0.16 ohms	3 V	-5%
	cathode: stainless steel					
2	anode: Al-covered stainless steel (Al purity: 99.99%)	phenolic active carbon fibers	3 F	0.16 ohms	3 V	-5%
	cathode: stainless steel					
3	anode: Al-covered stainless steel (Al purity: 99.99%)	phenolic active carbon fibers	3 F	0.16 ohms	3 V	-5%
	cathode: Al-covered stainless steel (Al purity: 99.86%)					

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 Capacitors of Invention

4 anode: Al-covered coconut 0.8 F 0.1 ohm 3 V -5%  
 stainless steel shell active  
 steel carbon particles  
 (Al purity:  
 99.86%)

cathode: stainless  
 steel

5 anode: Al-covered coconut 0.8 F 0.1 ohm 3 V -5%  
 stainless steel shell active  
 steel carbon particles  
 (Al purity:  
 99.86%)

cathode: Al-covered  
 stainless  
 steel  
 (Al purity:  
 99.99%)

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 Prior Art Capacitors

6 stainless steel phenolic 3 F 0.16 ohms 2.3 V -95%  
 for both anode active carbon  
 and cathode fibers

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## [Example 3]

As shown in Figs. 6 and 7, a polarizable electrode 6 of an acrylic active carbon fiber cloth (thickness 0.5 mm, specific surface area  $800 \text{ m}^2/\text{g}$ ) is formed with a 250 micrometer thick aluminium conductive electrode 7 by plasma spray coating. This double-layered construction is punched into a disk having a diameter of 2 cm, thereby obtaining an anode electrode. This electrode is superposed with a

lithium non-polarizable electrode 13 having a diameter of 2 cm through a separator 11 to obtain an electrode pair. This pair is impregnated with an electrolytic solution of 10 wt% of lithium tetrafluoroborate in propylene carbonate and sandwiched with stainless steel case pieces 9, 10 in which the stainless steel piece 9 alone or the pieces 9, 10 are covered with an aluminium layer (purity 99.99%, thickness 60 micrometers) on the inner surfaces thereof. The opening end between the case pieces 9, 10 is provided with a gasket 12 and caulked to close the pieces.

In Table 2, there are shown characteristics of the electric double layer capacitors of the invention as Nos. 1 and 2. For comparison, there are also shown characteristics of an electric double layer capacitor in which no aluminium layer is formed on the inner surfaces of a metallic case as No. 3.

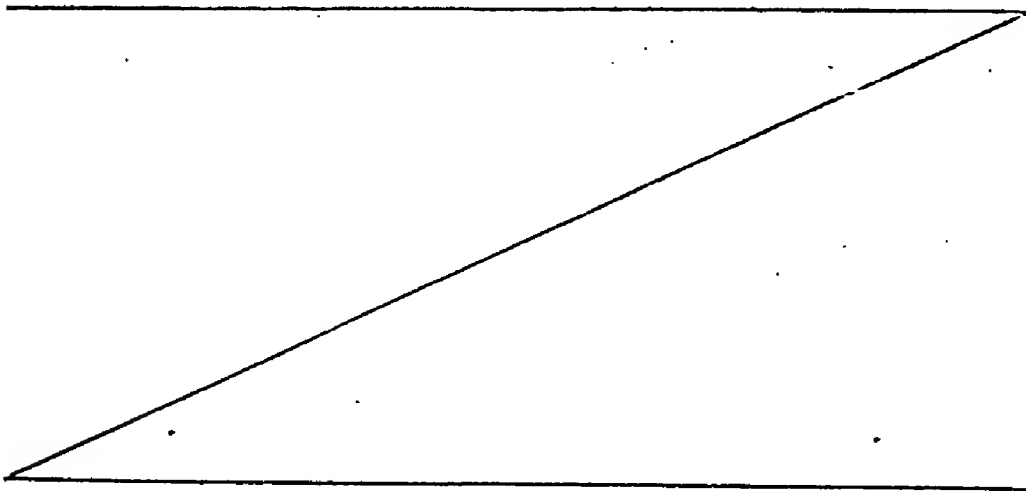


Table 2

No.	Case Construction	Combination of Electrodes	Capacitance	Internal Resistance	Break-down Voltage	Life under High Temperature and Loading Conditions
						3.0 V, 70°C 1000 hours
Capacitors of Invention:						(Variation in capacitance)
1	anode: Al-covered stainless steel (Al purity: 99.99%) cathode: stainless steel	facing type of a polarizable electrode & a non-polarizable electrode	6 F	0.15 ohms	3 V	-5%
2	anode: Al-covered stainless steel (Al purity: 99.99%) cathode: Al-covered stainless steel	.	6 F	0.15 ohms	3 V	-5%
Prior-art Capacitor:						
3	stainless steel for both an anode and a cathode	.	6 F	0.15 ohms	2.8 V	-50%

## INDUSTRIAL UTILITY

As will be understood from the foregoing, according to

the present invention, an electric double layer capacitor of a high breakdown voltage of 3 V or higher can be readily fabricated in which an aluminium layer is formed at least on the inner surface of a metallic case at the side of the anode. The aluminium layer is formed with an electrochemically stable anodized film according to an applied voltage and the resistance of the film is so low as not to present practical problems. This property of the aluminium film is effectively utilized for the fabrication.

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LIST OF REFERENCE NUMERALS OF THE DRAWINGS

- 1, 6 ... POLARIZABLE ELECTRODE
- 2, 9, 10 ... METALLIC CASE
- 3, 11 ... SEPARATOR
- 4, 12 ... GASKET
- 5 ... NON-POLARIZABLE ELECTRODE
- 7 ... CONDUCTIVE ELECTRODE
- 8 ... ALUMINIUM LAYER
- 13 ... COUNTER ELECTRODE

WHAT IS CLAIMED IS:

1. An electric double layer capacitor of the type which comprises a polarizable electrode having a conductive electrode on one side thereof, a counter electrode provided on the other side of said polarizable electrode through an electrolytic solution and a separator, and two metallic pieces for sealingly encasing said polarizable electrode and said counter electrode which are electrically interconnected with the two metallic pieces, respectively, at least a metallic piece provided at the side of the polarizable electrode being provided with an aluminium layer on the surface contacting the conductive electrode and the electrolytic solution.

2. An electric double layer capacitor according to Claim 1, wherein said counter electrode is a polarizable electrode of active carbon fibers or powder having a conductive electrode on one side thereof.

3. An electric double layer capacitor according to Claim 1, wherein said counter electrode is a non-polarizable electrode.

4. An electric double layer capacitor according to Claim 1, wherein said conductive electrode is made of aluminium.

5. An electric double layer capacitor according to Claim 5, wherein said metallic pieces are made of a member



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selected from the group consisting of stainless steel, iron, nickel, titanium, and copper alloys.

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FIG. 1

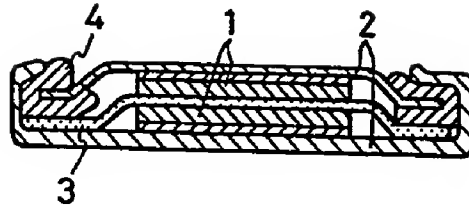


FIG. 2

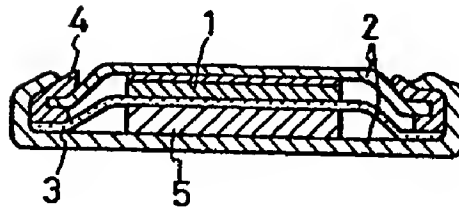
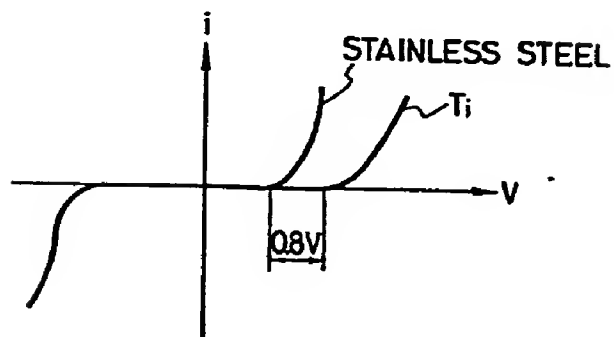


FIG. 3



2/2

FIG. 4

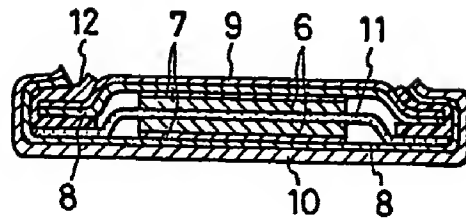


FIG. 5

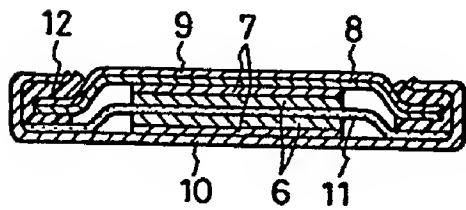


FIG. 6

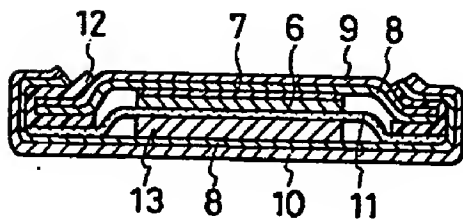
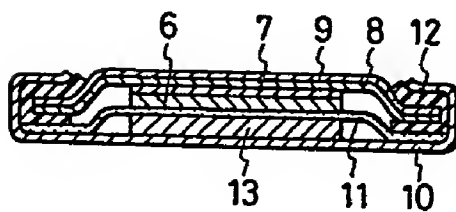


FIG. 7



## INTERNATIONAL SEARCH REPORT

International Application No. PCT/JP85/00706

I. CLASSIFICATION OF SUBJECT MATTER <small>If several classification symbols apply, indicate all *</small>		
According to International Patent Classification (IPC) or to both National Classification and IPC		
Int.Cl. <sup>4</sup> H01G 9/00		
II. FIELDS SEARCHED		
Minimum Documentation Searched *		
Classification System	Classification Symbols	
IPC	H01G 9/00	
Documentation Searched other than Minimum Documentation to the extent that such documents are included in the Fields Searched *		
Jitsuyo Shinan Koho 1926 - 1984 Kokai Jitsuyo Shinan Koho 1971 - 1984		
III. DOCUMENTS CONSIDERED TO BE RELEVANT *		
Category	Citation of Document ** with indication, where appropriate, of the relevant passages	Relevant to Claim No. <sup>13</sup>
Y	JP, A, 59-4114 (Matsushita Electric Industrial Co., Ltd.), 10 January 1984 (10. 01. 84) & WO, A1, 8400246 & JP, A, 59-48917 & JP, A, 59-67617 & EP, A1, 112923 & JP, A, 59-138327	1 - 5
Y	JP, A, 59-151414 (Matsushita Electric Industrial Co., Ltd.), 29 August 1984 (29. 08. 84) (Family: none)	1 - 5
Y	JP, A, 47-13577 (The Standard Oil Company), 13 July 1972 (13. 07. 72) & GB, A, 1315084 & FR, B1, 2119994 & SE, B, 382523 & JP, B4, 56-28007	3
Y	JP, U, 56-56646 (Malcon Electronics Co., Ltd.), 16 May 1981 (16. 05. 81) (Family: none)	5
<p>* Special categories of cited documents <sup>14</sup></p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"I" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance, the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance, the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art</p> <p>"S" document member of the same patent family</p>		
IV. CERTIFICATION		
Date of the Actual Completion of the International Search		Date of Making of this International Search Report
March 17, 1986 (17. 03. 86)		March 24, 1986 (24. 03. 86)
International Searching Authority		Signature of Authorized Officer
Japanese Patent Office		